

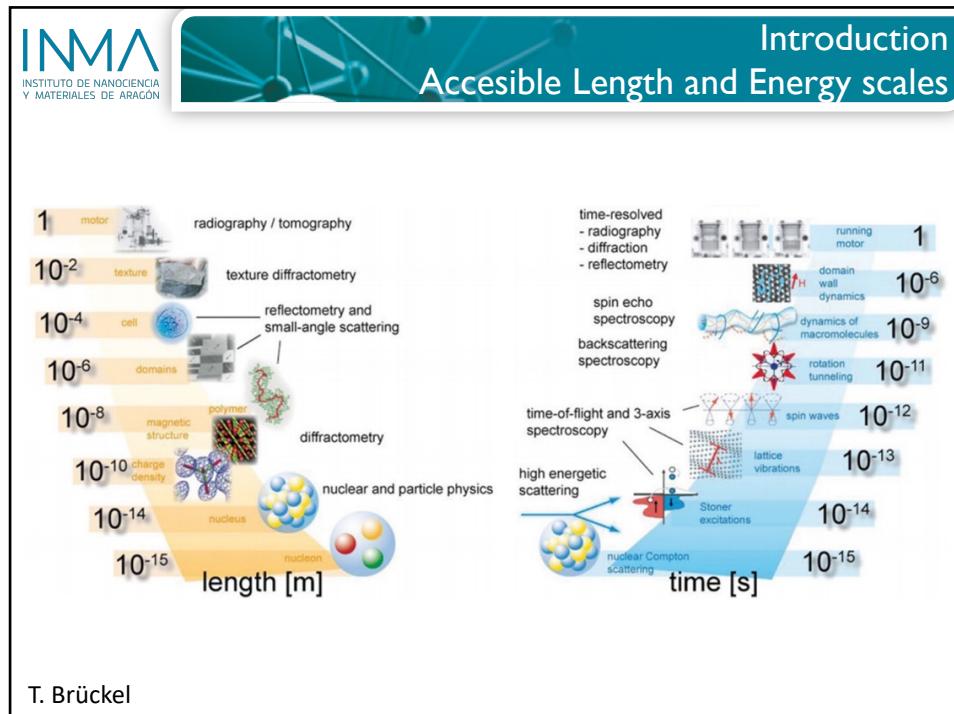


1

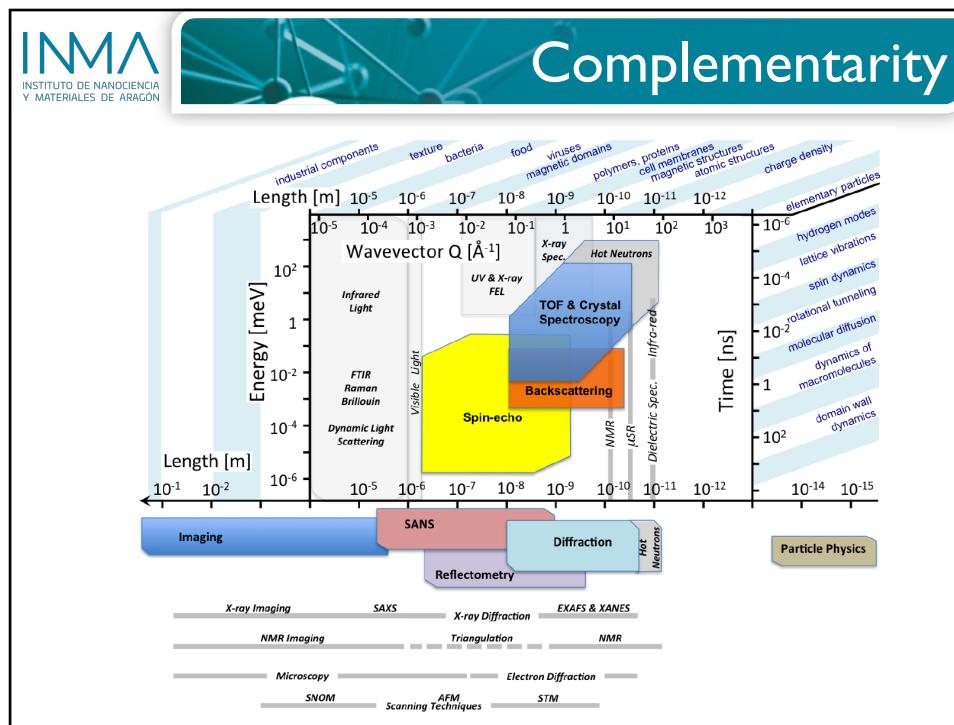
The outline slide has a teal header bar with the INMA logo and the word 'Outline'. The main content area is white and contains a bulleted list of topics:

- Introduction
- Basic concepts
- Instruments
- Examples
- How to apply for beam time

2



4



5

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Outline

- Introduction
- Basic concepts
- Instruments
- Examples
- How to apply for beam time

6

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Probing the matter

The diagram shows a cross-section of a material with atoms represented by circles containing dots. A red arrow labeled 'Neutron' enters from the left, interacting with the nuclei. A blue arrow labeled 'X Ray' also enters from the left, interacting with the electrons. A yellow arrow labeled 'Electron' enters from the bottom, interacting with the nuclei. The interactions are labeled: 'Nuclear Scattering' (red), 'Nuclear Interaction' (red), 'Dipole-dipole Interaction' (red), 'Electromagnetic Interaction' (blue), and 'Electrostatic Interaction' (yellow). The surface of the material is indicated on the left.

R Pynn

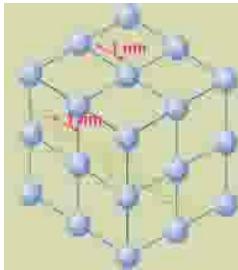
7

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

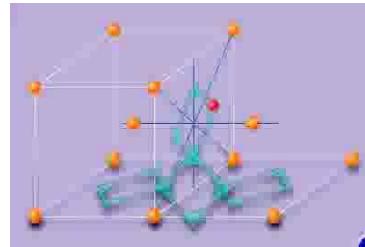
Basic concepts

Properties

- **Neutrons have similar wavelengths that the interatomic distances and therefore they can diffract**
 - Wide range of wavelengths (0.05 to 20 Å)
 - To observe different length scales



- **Same energy range that the atomic and electronic processes (meV a eV) with the possibility to detect changes less than μeV**
 - Tunneling,
 - Rotations
 - Vibrations
 - Electronic Transitions
 - Diffusion.



8

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

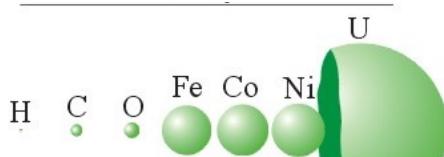
Basic concepts

Properties

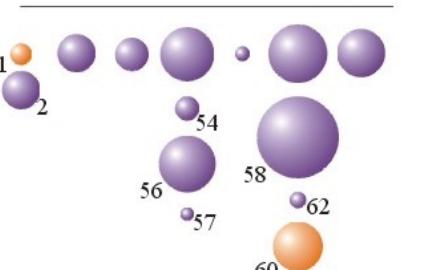
- **The interaction neutron-matter is via nucleus (**strong interaction**) and not with the electronic cloud**
 - Scattering length for atoms does not depend on the atomic number Z
 - Easy to see **light atoms** (H, Li) in presence of heavy atoms
 - Easy to distinguish **next neighbors** atoms in the periodic table

Atomic number	Name	Symbol	Atomic weight
1	H	H	1.008
2	He	He	4.003
3	Li	Li	6.941
4	Be	Be	9.012
5	B	B	10.811
6	C	C	12.011
7	N	N	14.012
8	O	O	15.999
9	F	F	18.998
10	Ne	Ne	20.180
11	Na	Na	22.990
12	Mg	Mg	24.310
13	Al	Al	26.982
14	Si	Si	28.085
15	P	P	30.973
16	S	S	32.065
17	Cl	Cl	35.453
18	Ar	Ar	39.902
19	K	K	39.098
20	Ca	Ca	40.080
21	Ti	Ti	46.000
22	V	V	50.942
23	Cr	Cr	51.996
24	Mn	Mn	54.938
25	Fe	Fe	55.845
26	Co	Co	58.933
27	Ni	Ni	58.693
28	Cu	Cu	63.546
29	Zn	Zn	65.401
30	Ga	Ga	69.724
31	Ge	Ge	72.610
32	As	As	74.920
33	Sb	Sb	75.000
34	Te	Te	78.900
35	I	I	74.920
36	At	At	82.630
37	Rn	Rn	85.460
38	Fr	Fr	87.860
39	Rb	Rb	85.460
40	Y	Y	88.905
41	Zr	Zr	91.224
42	Hf	Hf	178.490
43	Ta	Ta	180.947
44	W	W	183.840
45	Re	Re	186.200
46	Os	Os	190.230
47	Ir	Ir	192.210
48	Pt	Pt	195.080
49	Au	Au	196.970
50	Hg	Hg	200.590
51	Tl	Tl	204.380
52	Pb	Pb	207.200
53	Bi	Bi	208.980
54	Po	Po	210.000
55	A	A	222.000
56	Rn	Rn	222.000
57	Og	Og	226.000
58	Fr	Fr	223.000
59	Rb	Rb	224.000
60	Y	Y	227.000
61	La	La	231.000
62	Ce	Ce	232.000
63	Pr	Pr	233.000
64	Nd	Nd	234.000
65	Pm	Pm	235.000
66	Sm	Sm	237.000
67	Eu	Eu	238.000
68	Gd	Gd	239.000
69	Tb	Tb	240.000
70	Dy	Dy	241.000
71	Ho	Ho	242.000
72	Er	Er	243.000
73	Tm	Tm	244.000
74	Yb	Yb	247.000
75	Lu	Lu	248.000
76	Ac	Ac	247.000
77	Th	Th	261.000
78	Pa	Pa	261.000
79	U	U	261.000
80	Np	Np	261.000
81	Pu	Pu	261.000
82	Am	Am	261.000
83	Cm	Cm	261.000
84	Bk	Bk	261.000
85	Cf	Cf	261.000
86	Es	Es	261.000
87	Fm	Fm	261.000
88	Md	Md	261.000
89	No	No	261.000
90	Lw	Lw	261.000

X-rays form factors



Neutron Scattering length



9

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Basic concepts
Properties

The dependence of the scattering length with the nuclei do that different isotopes of the same element scatter in a very different way (**isotopic substitution and contrast experiments**)

10

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Basic concepts
Properties

Neutrons carry magnetic moment then they “see” (via the magnetic dipolar interaction) the magnetism present in the matter; unpaired electrons, nuclei

- To study magnetic structures
- Magnetic excitations

11

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Basic concepts
Properties

● With neutrons we can measure simultaneously the **structure** and the **dynamics**

Phonons

Magnons

12

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Basic concepts
Properties

Penetration. Neutron enters into materials without problems

- X Ray, EM , optical methods are surface probes.
- Hard exp. conditions
 - Cryostats
 - Furnaces
 - Pression clamps
 - Electrochemical cells
 - Etc...

Penetration depth (m)

Atomic number

Element	Neutrons (m)	X Rays (m)	Electrons (m)
Boron (B)	~0.0001	~0.0001	~0.0001
Calcium (Ca)	~0.01	~0.001	~0.0001
Iron (Fe)	~0.01	~0.001	~0.0001
Cobalt (Co)	~0.01	~0.001	~0.0001
Sodium (Na)	~0.01	~0.001	~0.0001
Chromium (Cr)	~0.01	~0.001	~0.0001
Cadmium (Cd)	~0.01	~0.001	~0.0001
Indium (In)	~0.01	~0.001	~0.0001
Samarium (Sm)	~0.01	~0.001	~0.0001
Potassium (K)	~0.01	~0.001	~0.0001
Lead (Pb)	~0.01	~0.001	~0.0001
Uranium (U)	~0.01	~0.001	~0.0001

13

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Basic Concepts
Neutron production: nuclear reactors

14

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Basic Concepts
Neutron production: spallation sources

- H^+ accelerated to $1\sim 5$ GeV
- Target W, Pb, Hg...
- Average 20 to 25 neutrons / H^+
- 50 to 16 Hz Pulses
- Pulse time $\sim \mu s$ to ms
- Pulsed Flux $3.7 \cdot 10^{16} \text{ ns}^{-1}$ (ISIS)

Proton très énergétique
Pb

- SNS @ ORNL
- ISIS @ United Kingdom
- MLF @ JPARC

15

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Basic Concepts
Neutron production: spallation sources



16

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Basic Concepts
Neutron production: spallation sources



17

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Basic Concepts
Neutron production: spallation sources



An aerial photograph showing the large circular ESS reactor building and its surrounding infrastructure, including roads, parking lots, and industrial buildings, situated near a body of water.

18

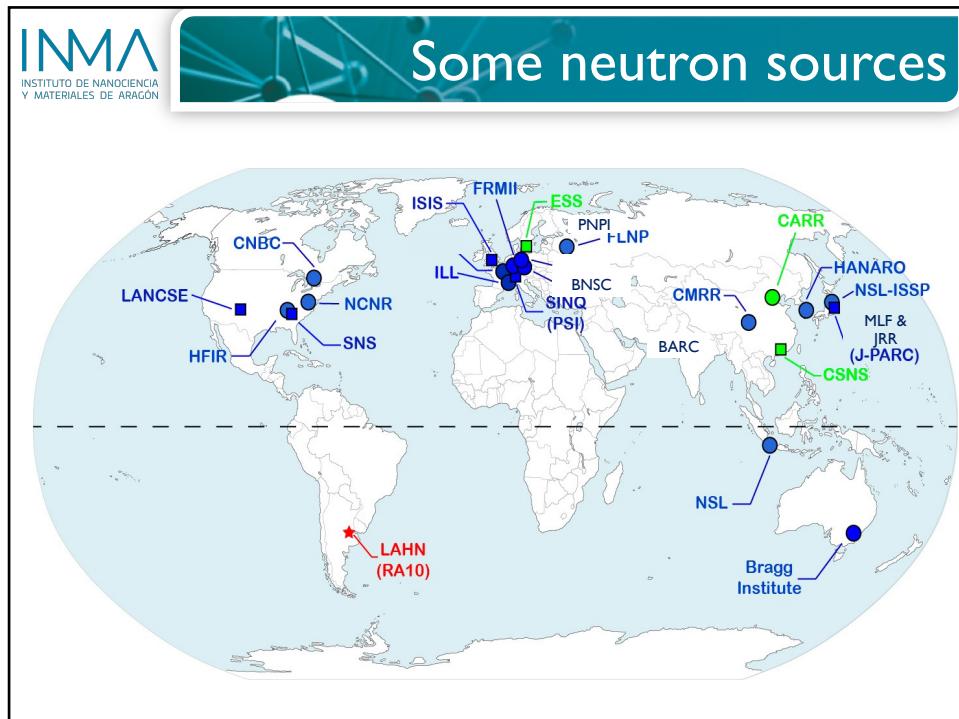
INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

The European spallation source, ESS

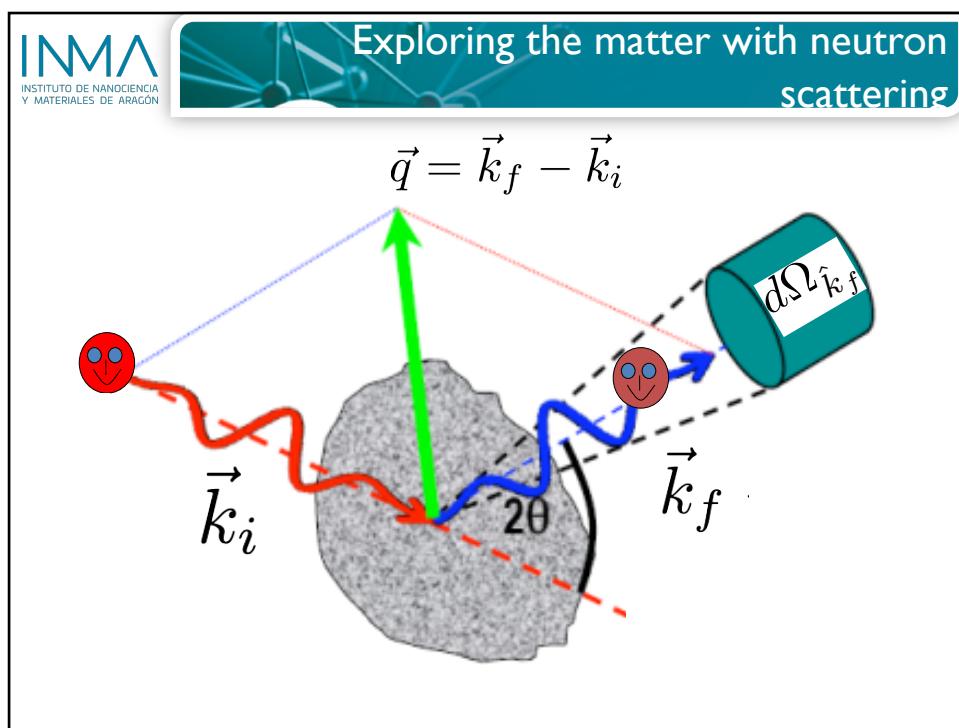


An aerial photograph showing the large circular ESS reactor building and its surrounding infrastructure, including roads, parking lots, and industrial buildings, situated near a body of water.

19



20



21

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Scattering cross section

- Double partial differential cross section

Number of scattered neutrons per second in a solid angle $d\Omega_{\hat{k}_f}$ in the direction given by \hat{k}_f with final energy between E_f and $E_f + dE_f$

$$\frac{d^2\sigma}{d\Omega_{\hat{k}_f} dE_f} = \frac{\varphi d\Omega_{\hat{k}_f} dE_f}{\text{Number of scattered neutrons per second}}$$

22

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Neutron scattering basic theory

- Consideremos un caso en el que la aproximación dipolar es válida y hagamos uso de las expresiones siguientes.

$$\delta(E_\lambda - E_{\lambda'} + \hbar\omega) = \frac{1}{2\pi\hbar} \int_{-\infty}^{+\infty} \exp\{i(E_{\lambda'} - E_\lambda)t/\hbar\} e^{-i\omega t} dt$$

$$\exp\{i\mathbf{H}t/\hbar\}|\lambda\rangle = \exp(iE_\lambda t)|\lambda\rangle$$

- Llegamos a

$$\sum_{\lambda\lambda'} p_\lambda \langle \lambda | \exp(-i\mathbf{k}\cdot\mathbf{R}_{ld'}) \mathbf{J}_{ld'}^\alpha | \lambda' \rangle \langle \lambda' | \exp(i\mathbf{k}\cdot\mathbf{R}_{ld}) \mathbf{J}_{ld}^\beta | \lambda \rangle \delta(E_\lambda - E_{\lambda'} + \hbar\omega)$$

$$= \frac{1}{2\pi\hbar} \int_{-\infty}^{+\infty} \langle \exp\{-i\mathbf{k}\cdot\mathbf{R}_{ld'}(t)\} \mathbf{J}_{ld'}^\alpha(0) \exp\{i\mathbf{k}\cdot\mathbf{R}_{ld}(t)\} \mathbf{J}_{ld}^\beta(t) \rangle \exp(-i\omega t) dt$$

- Y la sección amplitud

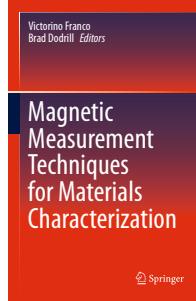
$$\left(\frac{d^2\sigma}{d\Omega dE} \right) = \frac{(\gamma r_0)^2}{2\pi\hbar} \frac{k'}{k} \sum_{\alpha\beta} (\delta_{\alpha\beta} - \hat{k}_\alpha \hat{k}_\beta) \sum_{ld'} \sum_{ld} \frac{e^{-i\mathbf{k}\cdot\mathbf{R}_{ld}}} {4\pi} F_{d'}^*(\mathbf{k}) F_d(\mathbf{k}) \times$$

$$\int_{-\infty}^{+\infty} \langle \exp\{-i\mathbf{k}\cdot\mathbf{R}_{ld'}(0)\} \exp\{i\mathbf{k}\cdot\mathbf{R}_{ld}(t)\} \rangle \langle \mathbf{J}_{ld'}^\alpha(0) \mathbf{J}_{ld}^\beta(t) \rangle \exp(-i\omega t) dt$$

23

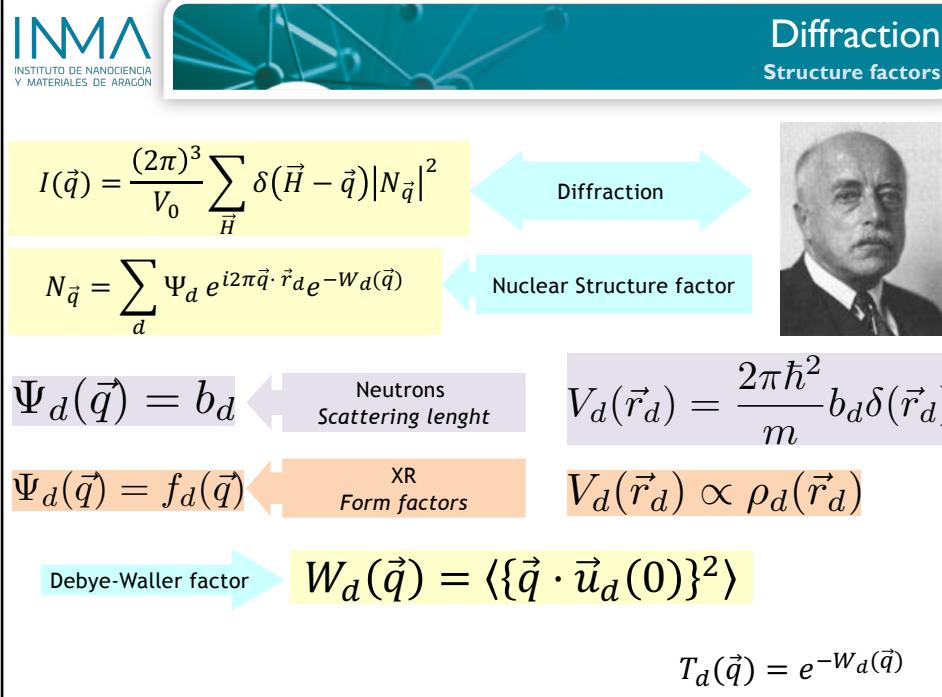
Bibliography

- G. L. Squires, “*Thermal neutron scattering*”, Cambridge University Press, 1978
- S.W. Lovesey, “*Theory of neutron scattering from Condensed Matter*”, Vol I y II Oxford University Press, 1986
- Javier Campo and Víctor Laliena “*Neutron Scattering in Magnetism: Fundamentals and Examples*” https://doi.org/10.1007/978-3-030-70443-8_14



24

Diffraction Structure factors



$$I(\vec{q}) = \frac{(2\pi)^3}{V_0} \sum_{\vec{H}} \delta(\vec{H} - \vec{q}) |N_{\vec{q}}|^2$$

$$N_{\vec{q}} = \sum_d \Psi_d e^{i 2\pi \vec{q} \cdot \vec{r}_d} e^{-W_d(\vec{q})}$$

$$\Psi_d(\vec{q}) = b_d$$

$$\Psi_d(\vec{q}) = f_d(\vec{q})$$

$$W_d(\vec{q}) = \langle \{\vec{q} \cdot \vec{u}_d(0)\}^2 \rangle$$

$$T_d(\vec{q}) = e^{-W_d(\vec{q})}$$

25

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

**General expression for NON polarised neutrons
in crystals**

Nuclear structure factor

$$\frac{d\sigma_{coh}^{(n, e)}}{d\Omega_{\hat{k}_f}} = \frac{N_c}{V} \sum_{\vec{H}} \delta(\vec{q} - \vec{H}) |N_{\vec{q}}|^2. \quad N_{\vec{q}} = \sum_d^{N_{uc}} b_d T_d(\vec{q}) e^{i2\pi\vec{q}\cdot\vec{r}_d}$$

Magnetic Interaction Vector

$$\frac{d\sigma_{coh}^{(m, e)}}{d\Omega_{\hat{k}_f}} = \frac{N_c}{V} \sum_{\vec{H}\vec{K}} \delta(\vec{q} - \vec{H} - \vec{K}) |\vec{M}_{\perp\vec{q}}|^2 \quad \vec{M}_{\perp\vec{q}} = \hat{q} \times (\vec{M}_{\vec{q}\vec{K}} \times \hat{q})$$

$$I_{\vec{q}} = |N_{\vec{q}}|^2 + |\vec{M}_{\perp\vec{q}}|^2$$

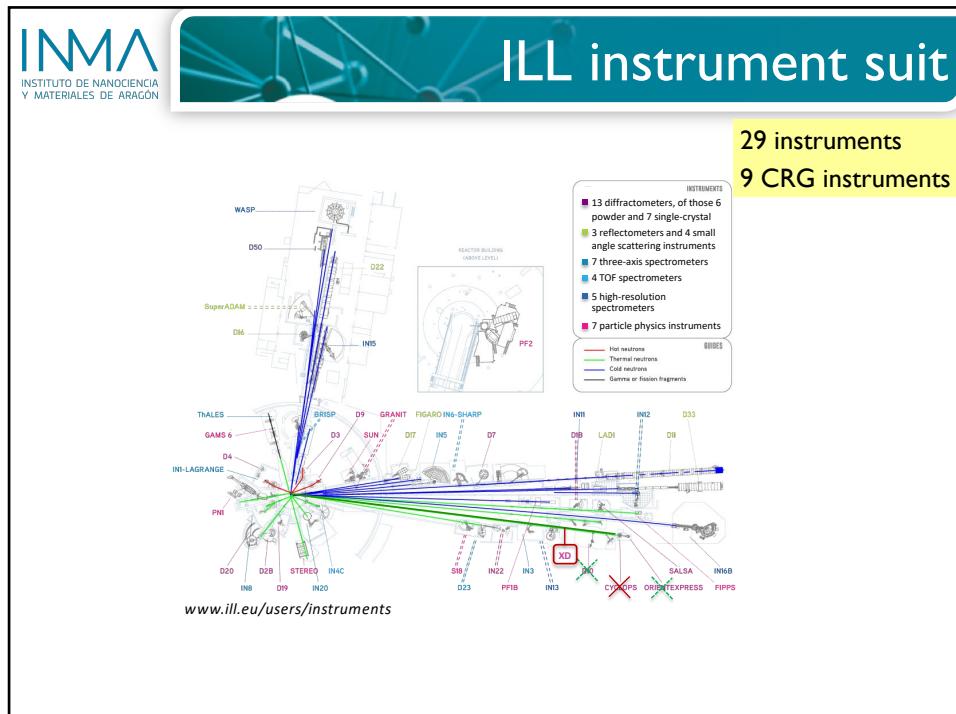
26

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Outline

- Introduction
- Basic concepts
- Instruments
- Examples
- How to apply for beam time

27



28



29

In brief




In brief



A **dedicated** diffractometer optimized for extreme conditions (**high P, high H, low and high T**) but with **flexibility** to accommodate a broad range of studies

Technically:
A flexible **powder diffractometer** with **single crystal capabilities**, with a variable **focused beam** on the sample, optimized signal/background ratio, **big solid angle** position-sensitive **detector**, and dedicated sample environment

30

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

What is around?

How XtremeD compares?

In the general diffraction context:

- Less P than X-ray but essential for **light elements**, contrast, 'high quality' crystallography and **magnetism**

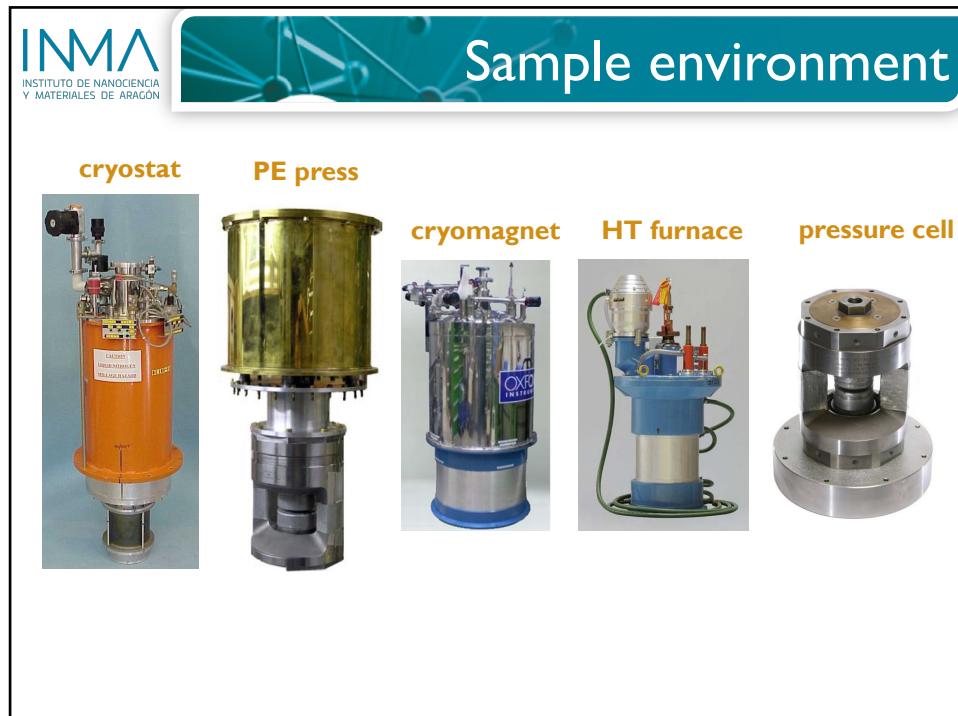
In the neutron facilities context:

- High intensity at low Q (w.r.t. most short pulse spallation instruments)
- Extended H & P range and **combination** of both parameters
- Both powders & single crystals
- Flexibility

In ILL's context:

- **Dedicated**
- Flux ~D20 but better signal to noise ratio and bigger solid angle
- Only single Xtal diffractometer at ILL with large 2D detector + high H (D19: space restrictions for high magnetic fields)

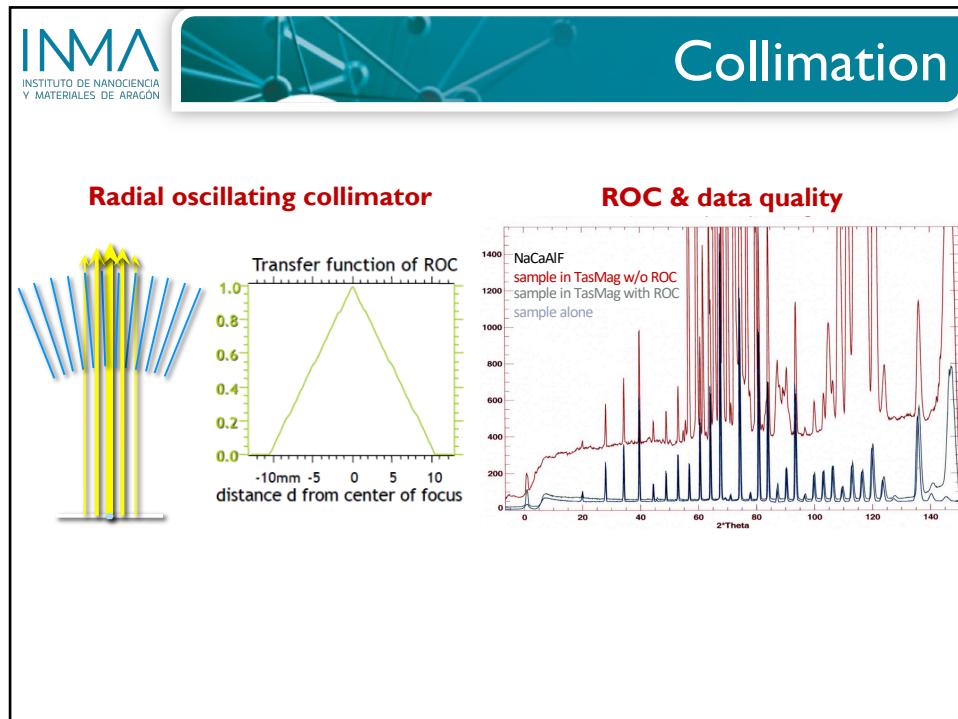
31



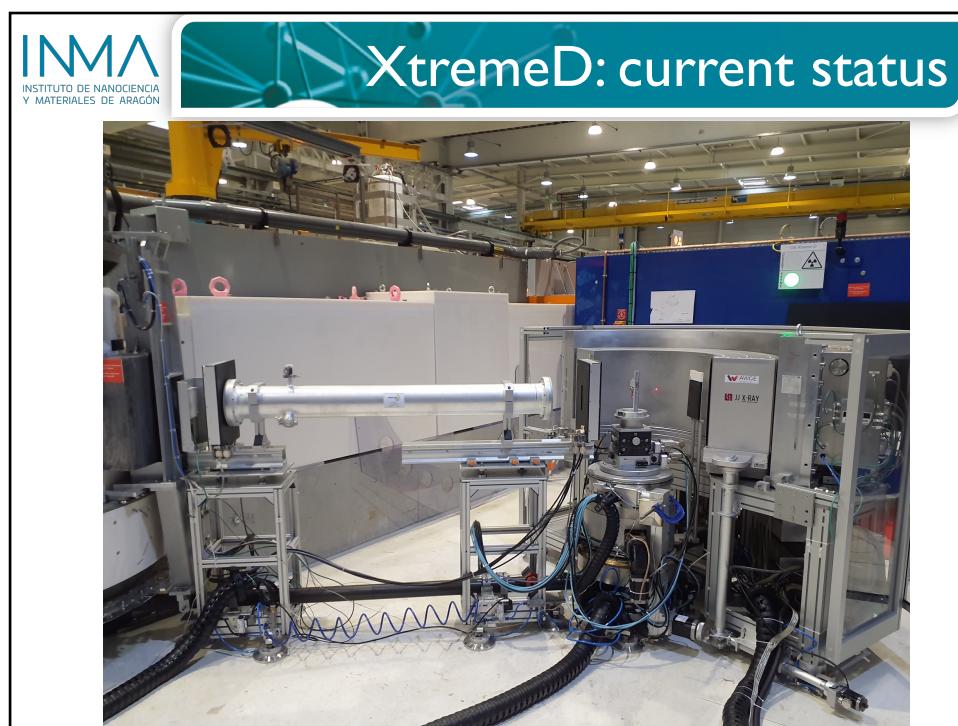
32



33



34



35

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

First neutrons

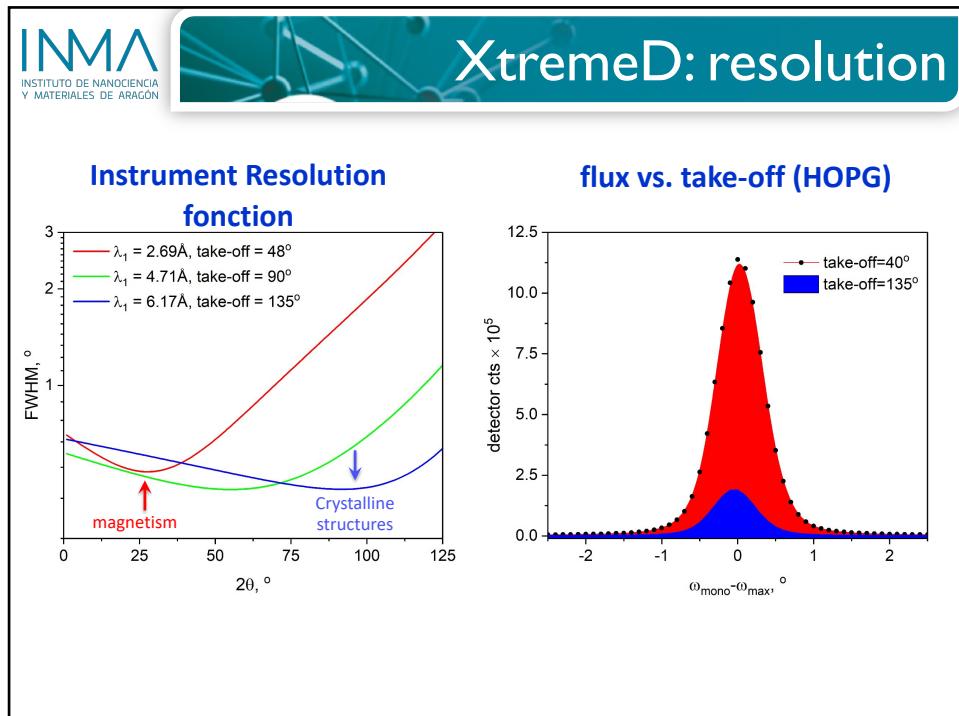
April 17, 2023

36

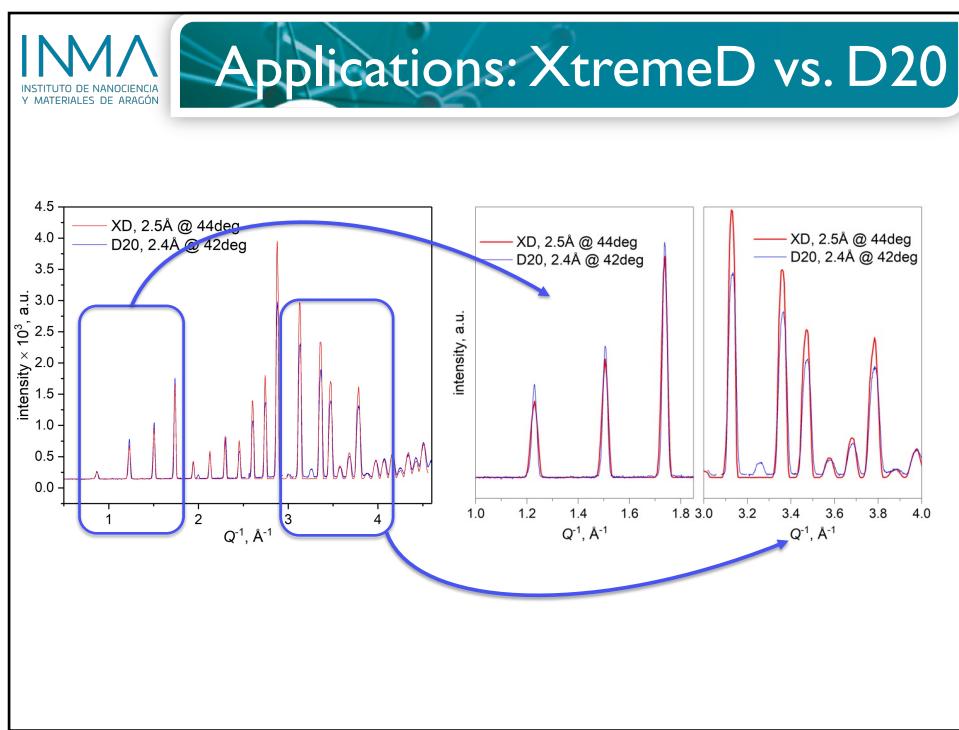
INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Powder diffraction

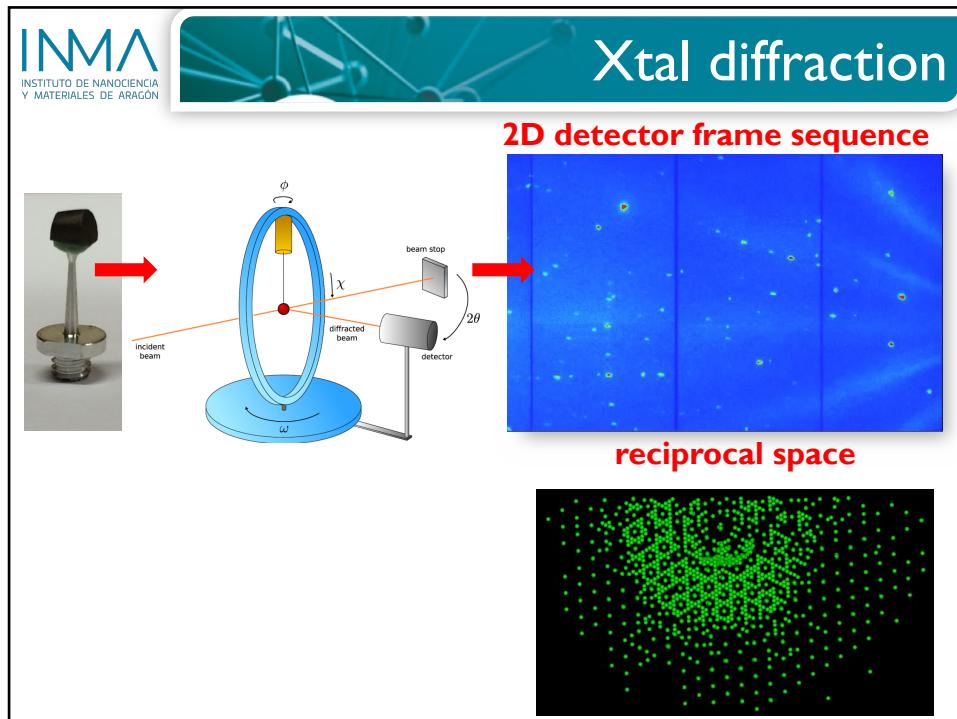
37



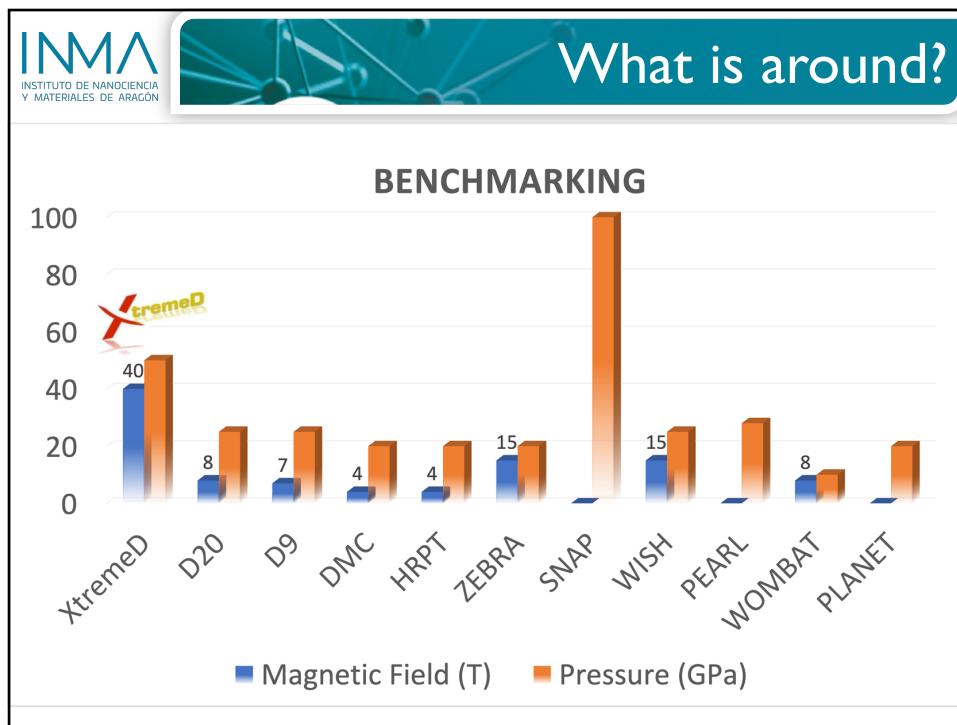
38



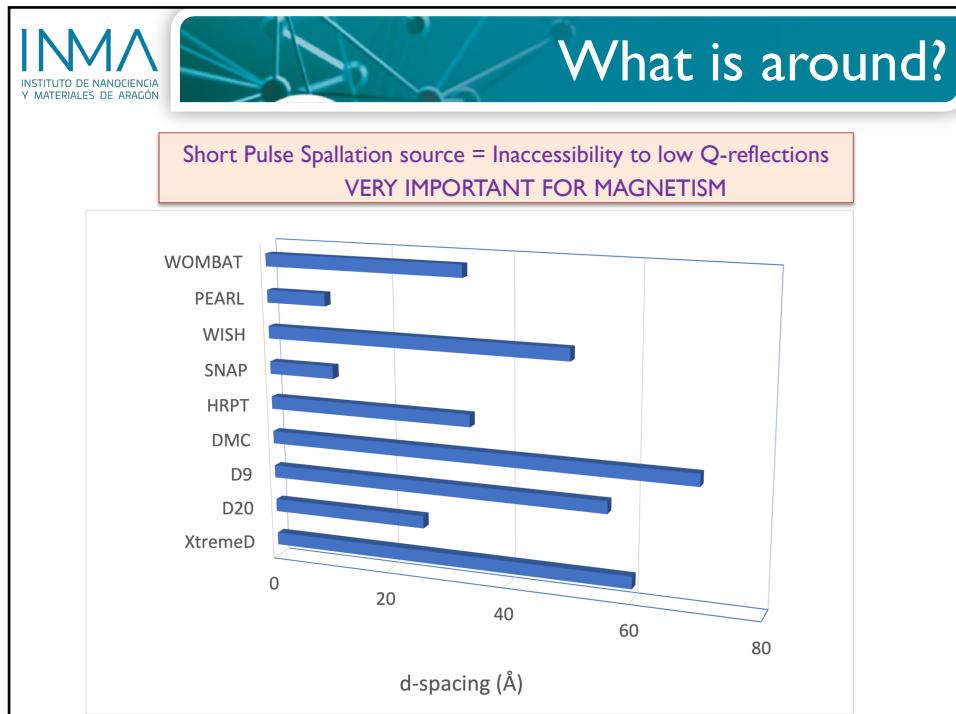
39



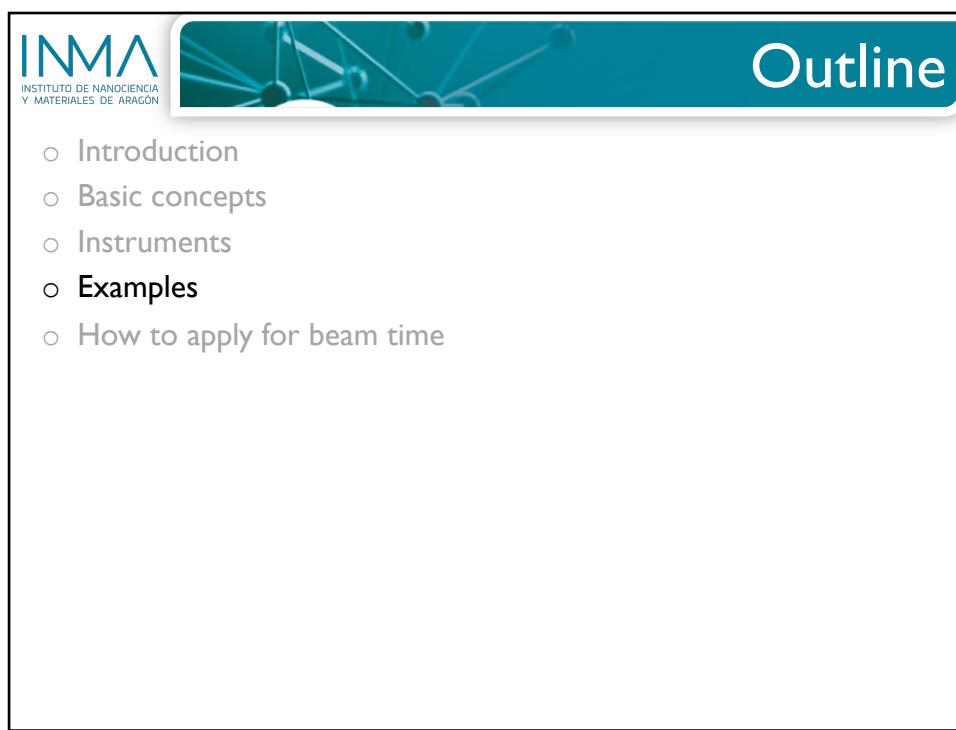
40



41



42



44

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Ringwoodite $[(\text{Mg},\text{Fe}^{2+})_2\text{SiO}_4 \text{ spinel}]$

research papers

Acta Cryst. B STRUCTURAL SCIENCE CRYSTAL ENGINEERING MATERIALS
ISSN 2052-5206

Determination of hydrogen site and occupancy in hydrous Mg_2SiO_4 spinel by single-crystal neutron diffraction

Narangoo Purevjav,^{a,*} Taku Okuchi,^a Xiaoping Wang,^b Christina Hoffmann^b and Naotaka Tomioka^c

Refined crystal structure of hydrous ringwoodite along [111] and hydrogen sites in a vacant Mg site.

The most plausible model (3H at 192*i* sites within an Mg-vacant octahedron)

45

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Kaolinite $[\text{Al}_2(\text{Si}_2\text{O}_5)(\text{OD})_4]$

$\text{Al}_2(\text{Si}_2\text{O}_5)(\text{OD})_4$

Crystal structure of kaolinite as refined from neutron diffraction. Layer slab with the $[\text{AlO}_6]$ -octahedra in dark shading showing hydroxyl terminations between layers (left) and the excess hydroxyl between tetrahedral silicate layer and those oxygens that are not charge balanced (right) are shown.

Neutrons clearly identify the location of the protons

E.Akiba, et al. Clays Clay Miner. 45, 781 (1997)

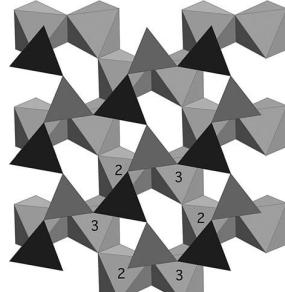
46

Cation order-disorder

- Cation order-disorder in Olivines and Spinel at high temperature
 - Fe-Ni, Mn-Fe, Mg-Al, Ni-Mn, Ni-Mg, etc...

- Mg/Al Ordering in Dioctahedral Micas
 - $K_{0.9}[Mg_{0.58}Al_{1.43}][Si_{3.57}Al_{0.43}]O_{10}(OH)_2$

- Al/Si and Mg/Al ordering in high-temperature amphibole
 - $NaCa_2[Mg_4Al][Si_6Al_2]O_{22}(OH)_2$



47

other

- Mechanical Behavior of Geological Materials
 - if the deformation experiment is performed on a neutron beam line, the deformation behavior within the sample can be monitored as it is being deformed from diffraction patterns collected at different stages of the deformation

- Texture determination
 - neutron texture goniometry affords true volume texture measurements of relatively large (up to several cm³) isometric samples

48

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

High Pressure

What would be the *neutrons at HP* contribution?

The study of these systems implies systematic studies of **hydrogen bonding and clustering, host-guest interactions in clathrate hydrates, the role of water in crystals and cation order disorder phase transitions.**

- Planetary ices (water, methane, ammonia) and clathrates
- Segregation and related phenomena in water solutions at extreme conditions
- Mineral hydration: implications for the water cycle in Earth crust and mantle
- Hydrothermal reactions
- Cation order-disorder

49

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Simple molecules	D ₂ O, D ₂ , NH ₃ , CH ₄
Larger molecules	KDP
Hydrides	MgH ₂ , gamma-CoH, MgD ₂ : TiD ₂ mixture, Fe hydrides, LiD, NaD, AlD ₃
perovskite hydrides	Na _{1-x} Li _x MgH ₃ , the ternary hydrides Mg ₂ NiH ₄ hydride and Mg ₃ CuH _x , Laves phase hydrides
Hydroxides	MOH family (M = Li, Na, K, ...), M(OH) ₂ hydroxides (M = Mn, Fe, Co, Ni, Cd, Mg and Ca)

Currently 16 distinct crystalline phases of ice have been measured experimentally.

Guthrie, J. Phys.: Condens. Matter 27 (2015) 153201
Pruzan P 1998 The Phase Diagram of H₂O

50

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Hydrogen atoms

PNAS Neutron diffraction observations of interstitial protons in dense ice

Malcolm Guthrie^{a,1}, Reinhard Boehler^a, Christopher A. Tulk^b, Jamie J. Molaison^b, António M. dos Santos^b, Kuo Li^a, and Russell J. Hemley^{a,1}

Location of hydrogen atoms

25-50 GPa
RT

Dense ices:

- Starting of the destabilisation of water molecule.
- This phase, precursor of superionic behaviour (ice X)?
- Intermediate dissociation, with **H occupying of interstitial sites**

51

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Phase transformations

Static Compression and H Disorder in Brucite, $Mg(OH)_2$, to 11 GPa: a Powder Neutron Diffraction Study

M. Catti¹, G. Ferraris², S. Hull³, A. Pavese⁴

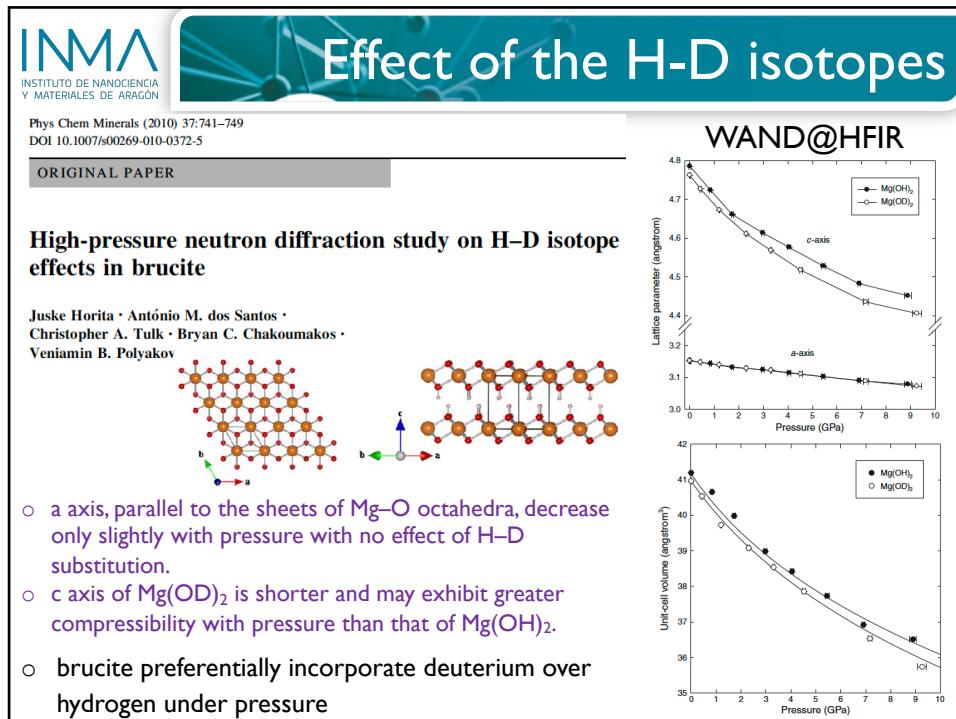
PHYSICS AND CHEMISTRY OF MINERALS
© Springer-Verlag 1995

Polaris @ISIS

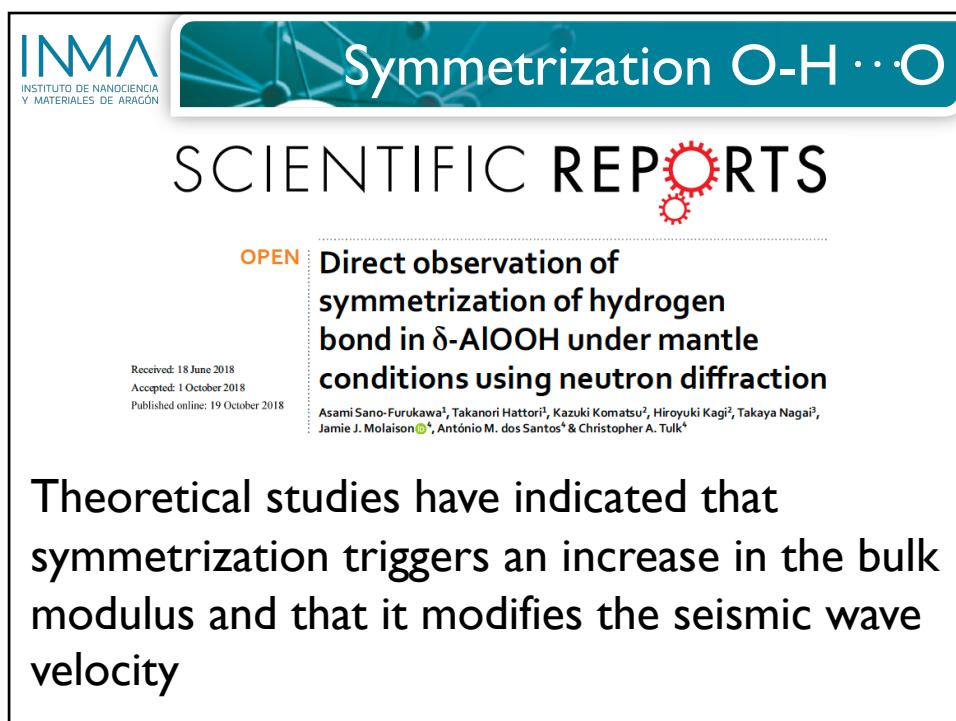
The onset of H disorder, and a jump of the c/a ratio vs. pressure at ~6.5 GPa, may be related to a second-order phase transition consistent with Raman results.

11 GPa
~6.5 GPa

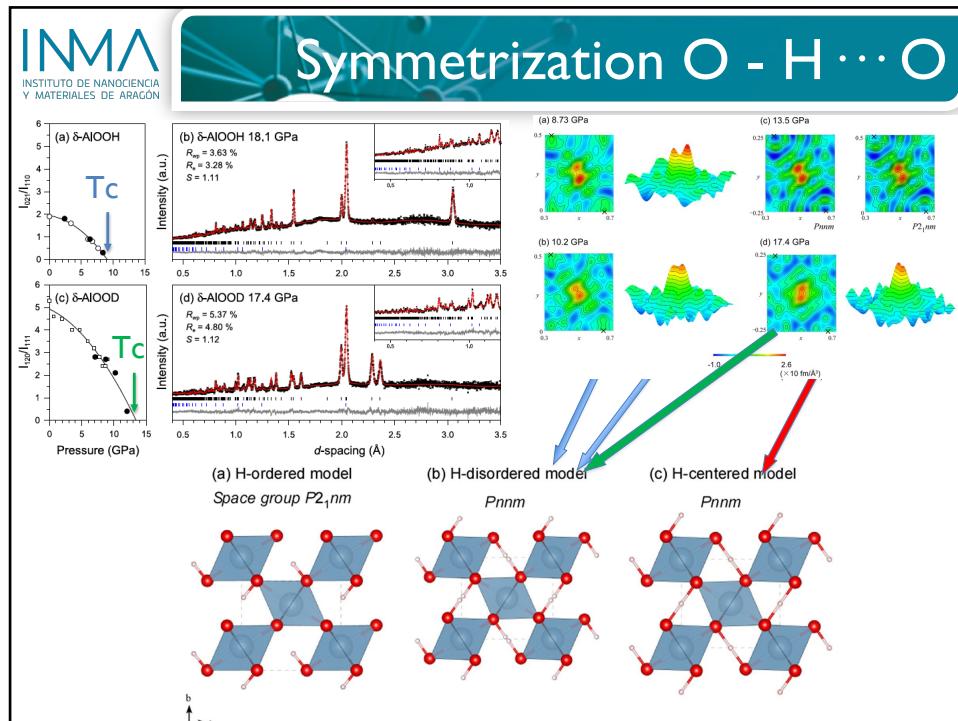
52



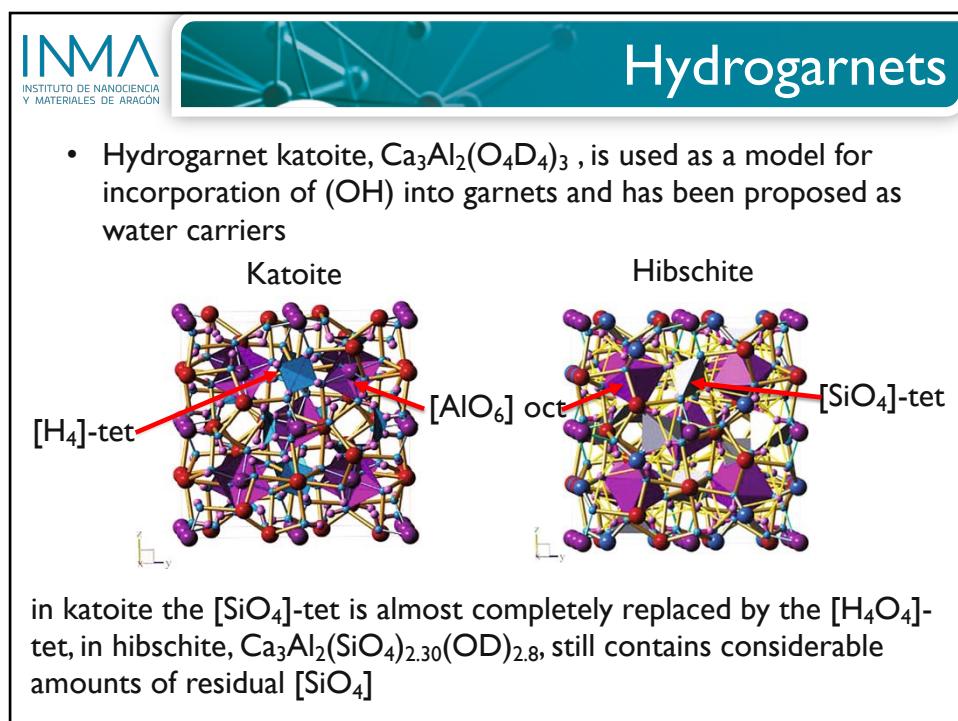
53



54



55



56

INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

High Pressure and Temperature

Physics and Chemistry of Minerals (2019) 46:459–469
<https://doi.org/10.1007/s00269-018-1016-4>

ORIGINAL PAPER

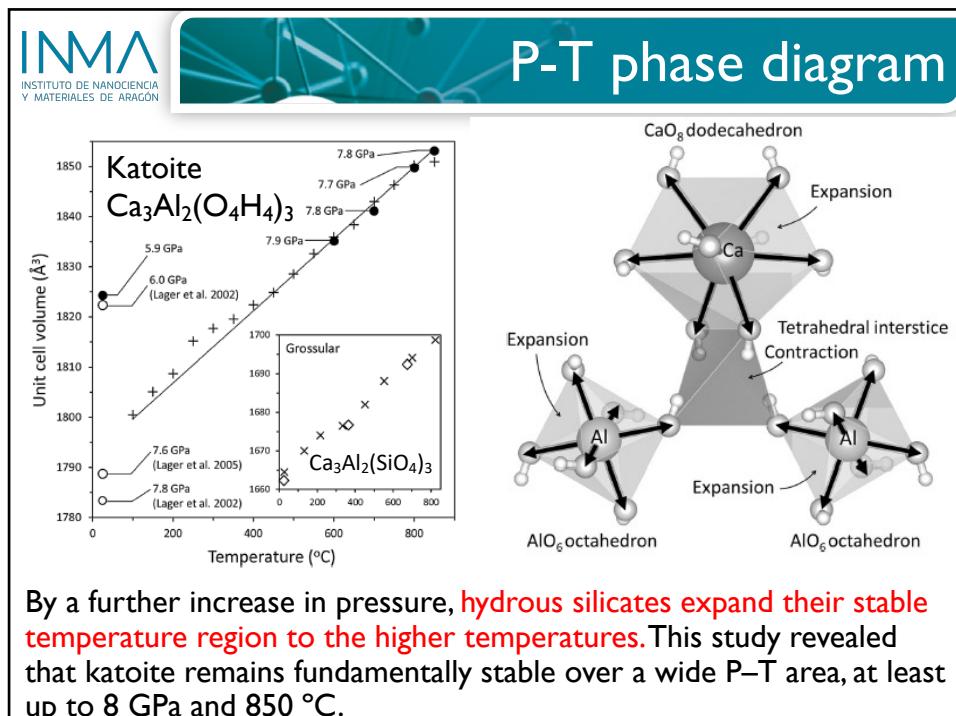
Crystal structure change of katoite, $\text{Ca}_3\text{Al}_2(\text{O}_4\text{D}_4)_3$, with temperature at high pressure

Atsushi Kyono¹ · Masato Kato¹ · Asami Sano-Furukawa² · Shin-Ichi Machida³ · Takanori Hattori²

effect of hydrogen on the katoite's phase transition under high-pressure and high-temperature conditions,
→ → high-temperature neutron diffraction measurements at about 8 GPa where it is expected that the phase transition from $Ia3d$ to another one is observed

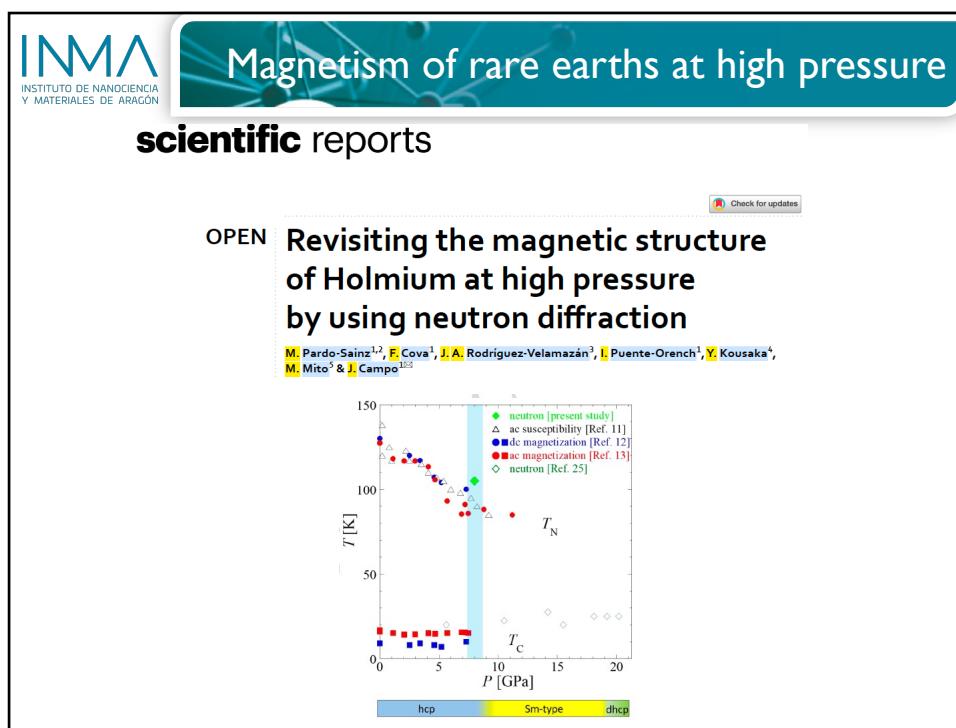
57

58

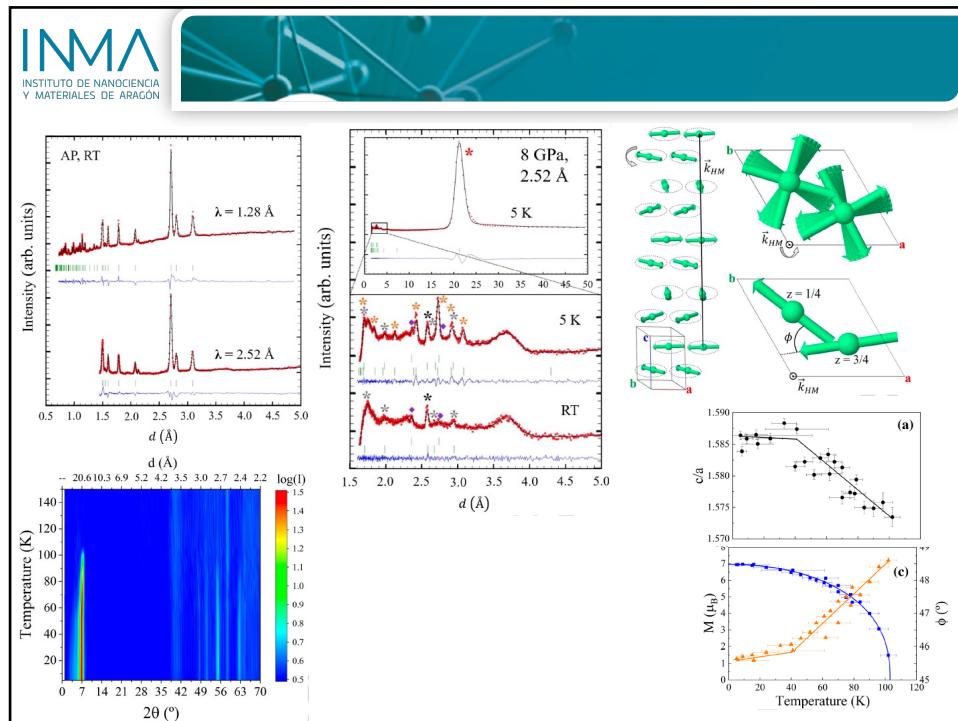


By a further increase in pressure, **hydrous silicates expand their stable temperature region to the higher temperatures**. This study revealed that katoite remains fundamentally stable over a wide P-T area, at least up to 8 GPa and 850 °C.

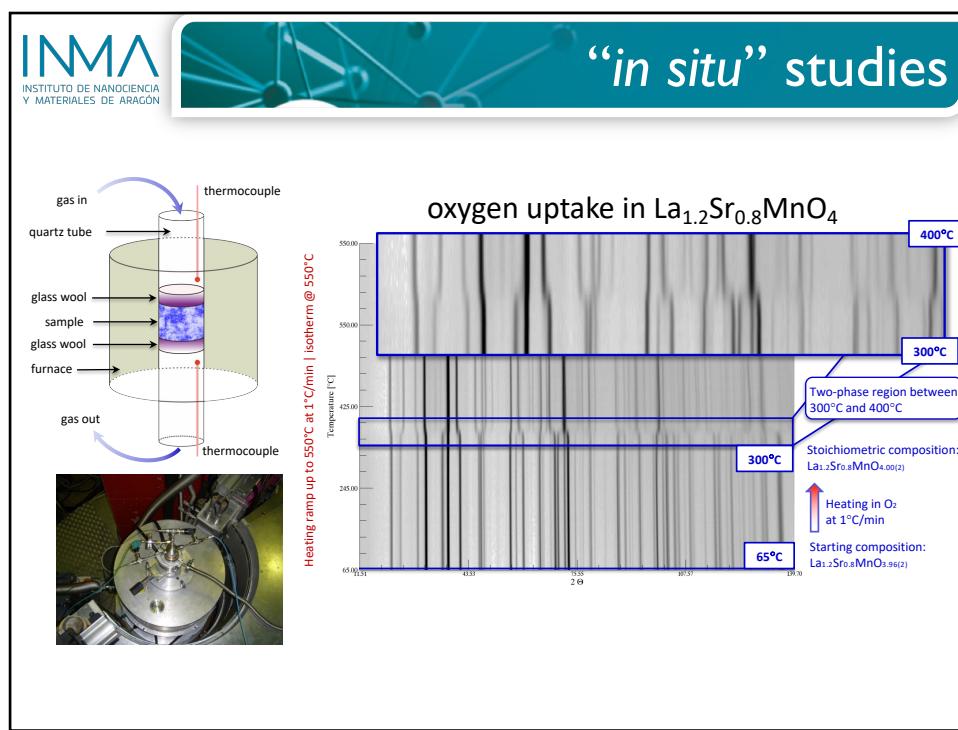
59



60



61



63



Outline

- Introduction
- Basic concepts
- Instruments
- Examples
- How to apply for beam time

64



How to apply for beamtime

- Do you have a good idea for neutron scattering?
- Is neutron scattering the only technique to solve the problem?
- Do you have any previous experiments relevant for the proposal?
- Did you contact any scientist specialist in NS to help you with the more technical questions? (Local Contact...)
- Did you think in the expected results of the experiment?
- How will you analyze your data?
- How did you estimate the requested beamtime?
- Which is the best instrument and neutron source?
- Etc...

65



INMA
INSTITUTO DE NANOCIENCIA
Y MATERIALES DE ARAGÓN

Thanks

Thank you very much for your attention !!

Our webpage
m4.unizar.es



MINISTERIO
DE CIENCIA
E INNOVACIÓN

AGENCIA
ESTATAL DE
INVESTIGACIÓN

CSIC
CONSEJO SUPERIOR DE INVESTIGACIONES CIENTÍFICAS

Universidad
Zaragoza